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# Groups and Representations

Instruction 8 for the preparation of the lecture on 17 May 2021

### **3** Applications in quantum mechanics

## 3.1 Expansion in irreducible basis functions and selections rules Setting:

 $L^2$ -spaces & unitary operators https://youtu.be/5yJbnMbWZK4 (2min) (1)

**Lemma 8.** Let G be a (finite) group of unitary linear operators  $V \to V$ ,  $A \in G^{1}$ , and let  $\psi_1^{\nu}, \ldots, \psi_{d_{\nu}}^{\nu}$  be functions that transform in the unitary irreducible representation  $\Gamma^{\nu}$  (with  $\dim(\Gamma^{\nu}) = d_{\nu}, \ i.e.$ 

$$A\psi^{\nu}_{\alpha} = \sum_{\beta=1}^{d_{\nu}} \psi^{\nu}_{\beta} \, \Gamma^{\nu}(A)_{\beta\alpha} \, .$$

Then  $\exists C_{\nu} \in \mathbb{C}$  such that  $\langle \psi_{\alpha}^{\nu} | \psi_{\beta}^{\mu} \rangle = C_{\nu} \, \delta_{\nu\mu} \, \delta_{\alpha\beta}.$ 

**Proof:** 

$$https://youtu.be/Ru30m0wi0TM (7min)$$
(2)

**Remarks**:

Have you heard the term *selection rule* before? If not, never mind. If yes, in which context? Let's speak about it in the live session.

### 3.2 Invariance of the Hamiltonian and degeneracies

A special role is played by the Hamiltonian  $H: V \to V$  (a linear self-adjoint operator) of a quantum mechanical system. In particular, its eigenvalues are the possible energy levels in which we can find the system.

Let H be the Hamiltonian of a quantum mechanical system and let A be a unitary operator. If

$$AH = HA$$
,

we say that A commutes with the Hamiltonian or that A leaves H invariant.

The set of all symmetry operations (realised by unitary operators) that leave H invariant forms a group G, the symmetry group of H. Why is this a group?

Every eigenspace of H, say  $\{\psi \in V : H\psi = E\psi\}$ , carries a representation of the symmetry group G:

> (4)https://youtu.be/t0\_HAXTVgmg (2min)

<sup>&</sup>lt;sup>1</sup>Alternatively, view the operators A as unitary representation of a group G on V.

This representation can, in principle, be reducible or irreducible; typically it is irreducible:

 $\blacktriangleright$  All states transforming in the same irrep of G must have the same energy:

$$https://youtu.be/hdODKxaR4Ec (3 min)$$
(5)

► States transforming in different irreps can have different energies – at least, symmetry does not force them to have the same energy:

$$\texttt{https://youtu.be/er1ZQ3a8_38} (2\min) \tag{6}$$

If states transforming in different irreps still have the same energy, we speak about "accidental degeneracy". Possible reasons:

- $\blacktriangleright$  "Fine-tuning" of one or several parameters in H (unlikely).
- ▶ We haven't correctly identified the full symmetry group, i.e. we have overlooked some symmetry.

**Conclusions:** 

- Degenerate states to a given energy typically transform in an irrep of the symmetry group of H, i.e. they can be classified by irreps.
- $\blacktriangleright$  number of degenerate states = dimension of the irrep

#### Example (& outlook): Hydrogen atom